

catena-Poly[[(μ -4,4'-bipyridine- κ^2 N:N')-bis[aqua(dimethylformamide- κ O)-copper(II)]]-di- μ -terephthalato- κ^4 O¹:O⁴]

 Ji-Yang Xu,^a Bang-Lin Chen^{b*} and Seik Weng Ng^c

^aDepartment of Biochemistry, China Pharmaceutical University, Nanjing 210009, People's Republic of China, ^bDepartment of Chemistry, University of Texas – Pan American, Edinburg, TX 78539-2999, USA, and ^cDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia
Correspondence e-mail: banglin@utpa.edu

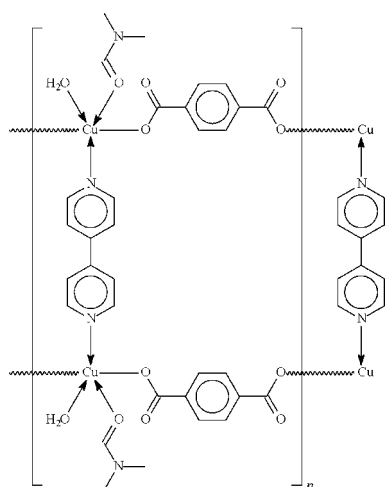
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.033; wR factor = 0.086; data-to-parameter ratio = 18.7.

The title coordination polymer, $[\text{Cu}_2(\text{C}_8\text{H}_4\text{O}_4)_2(\text{C}_{10}\text{H}_8\text{N}_2)(\text{C}_3\text{H}_7\text{NO})_2(\text{H}_2\text{O})_2]_n$, adopts a ladder structure in which terephthalate functions as the rails and bipyridine, lying on inversion centres, as the rungs. The Cu atom is also coordinated by water and dimethylformamide molecules in a square-pyramidal environment. Hydrogen bonds link the ladders into a three-dimensional network.

Related literature

For the crystal structure of a copper terephthalate-4,4'-bipyridine cocrystal with terephthalic acid, see Baeg & Lee (2003).



Experimental

Crystal data

 $[\text{Cu}_2(\text{C}_8\text{H}_4\text{O}_4)_2(\text{C}_{10}\text{H}_8\text{N}_2)(\text{C}_3\text{H}_7\text{NO})_2(\text{H}_2\text{O})_2]$
 $M_r = 793.71$
Monoclinic, $P2_1/n$
 $a = 10.8424$ (2) Å
 $b = 14.3465$ (3) Å
 $c = 11.1929$ (2) Å
 $\beta = 108.741$ (1)°
 $V = 1648.75$ (5) Å³
 $Z = 2$
Mo $K\alpha$ radiation
 $\mu = 1.36$ mm⁻¹
 $T = 173$ (2) K
 $0.15 \times 0.03 \times 0.02$ mm

Data collection

 Bruker X8 APEXII diffractometer
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
 $T_{\min} = 0.813$, $T_{\max} = 0.973$

 21007 measured reflections
4415 independent reflections
3131 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.043$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.033$
 $wR(F^2) = 0.086$
 $S = 1.04$
4415 reflections
236 parameters
2 restraints

 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.47$ e Å⁻³
 $\Delta\rho_{\min} = -0.45$ e Å⁻³
Table 1

Selected bond lengths (Å).

Cu1—O1	1.943 (1)	Cu1—O1 _w	1.975 (2)
Cu1—O4 ⁱ	1.946 (1)	Cu1—N1	2.008 (2)
Cu1—O5	2.312 (1)		

 Symmetry code: (i) $x + 1, y, z$.

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1W—H1W1 \cdots O2 ⁱⁱ	0.83 (1)	1.83 (1)	2.658 (2)	175 (3)
O1W—H1W2 \cdots O3 ⁱⁱⁱ	0.84 (1)	1.85 (1)	2.684 (2)	168 (3)

 Symmetry codes: (ii) $-x + 1, -y + 1, -z + 1$; (iii) $-x, -y + 1, -z + 1$.

Data collection: APEX2 (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2007).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CF2124).

References

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Westrip, S. P. (2007). publCIF. In preparation.

supplementary materials

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***catena*-Poly[[$(\mu$ -4,4'-bipyridine- $\kappa^2 N:N')$ bis[aqua(dimethylformamide- κO)copper(II)]]-di- μ -terephthalato- $\kappa^4 O^1:O^4$]**

J.-Y. Xu, B.-L. Chen and S. W. Ng

Comment

The hydrothermal reaction of copper(II) nitrate, terephthalic acid and 4,4'-bipyridine furnishes the expected copper terephthalate adduct with 4,4'-bipyridine as a 1/1 cocrystal with terephthalic acid. The compound adopts a layer structure and the terephthalic acid behaves as a guest molecule in the porous host (Baeg & Lee, 2003). With the addition of DMF in the hydrothermal synthesis, the dimensionality of the product is lowered to a ladder structure in the present study. The compound is formally $(C_{10}H_8N_2)(C_8H_4O_4)_2(C_3H_7NO)_2(H_2O)_2Cu_2$; the terephthalate represents the rails of the ladder and the *N*-heterocycle the rungs. The copper atom is also coordinated by water and DMF in a square-pyramidal geometry. Although the ladder appears to have voids (Fig. 2), these are actually occupied by the DMF ligands of adjacent ladders, and there are no empty spaces in the crystal structure.

Experimental

Copper(II) nitrate 2.5-hydrate (0.025 g, 0.11 mmol), terephthalic acid (0.018 g, 0.11 mmol) and 4,4'-bipyridine (0.009 g, 0.06 mmol) in DMF/ethanol/water (3 ml/3 ml/2 ml) were heated at 348 K for several days. Small, dark blue crystals were collected from the cooled solution in 70% yield.

Refinement

Carbon-bound hydrogen atoms were placed at calculated positions ($C-H = 0.95-0.98 \text{ \AA}$) and were included in the refinement in the riding model approximation, with $U_{iso}(H)$ set to 1.2–1.5 times $U_{eq}(C)$. The water H atoms were located in a difference Fourier map and were refined with a distance restraint of $O-H = 0.84 (1) \text{ \AA}$; the displacement parameters were freely refined.

Figures

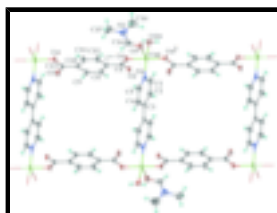


Fig. 1. A portion of the ladder structure; displacement ellipsoids are drawn at the 70% probability level, and H atoms as spheres of arbitrary radius. [Translational code (i): $x, 1 + y, z$.]

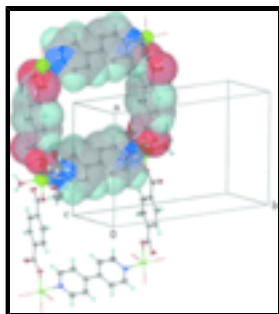


Fig. 2. Space-filling plot depicting the square grid formed from the bipyridine and terephthalate linkages that make up the ladder structure.

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Crystal data

$[\text{Cu}_2(\text{C}_8\text{H}_4\text{O}_4)_2(\text{C}_{10}\text{H}_8\text{N}_2)(\text{C}_3\text{H}_7\text{NO})_2(\text{H}_2\text{O})_2]$

$M_r = 793.71$

Monoclinic, $P2_1/n$

Hall symbol: -P 2yn

$a = 10.8424$ (2) Å

$b = 14.3465$ (3) Å

$c = 11.1929$ (2) Å

$\beta = 108.741$ (1)°

$V = 1648.75$ (5) Å³

$Z = 2$

$F_{000} = 816$

$D_x = 1.599$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 4717 reflections

$\theta = 2.4$ – 27.3 °

$\mu = 1.36$ mm⁻¹

$T = 173$ (2) K

Prism, blue

$0.15 \times 0.03 \times 0.02$ mm

Data collection

Bruker X8 APEXII
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 173$ (2) K

φ and ω scans

Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)

$T_{\min} = 0.813$, $T_{\max} = 0.973$

21007 measured reflections

4415 independent reflections

3131 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.043$

$\theta_{\max} = 29.4$ °

$\theta_{\min} = 2.3$ °

$h = -12 \rightarrow 14$

$k = -19 \rightarrow 19$

$l = -15 \rightarrow 15$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.033$

$wR(F^2) = 0.086$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0434P)^2 + 0.0236P]$

$S = 1.04$

4415 reflections

236 parameters

2 restraints

Primary atom site location: structure-invariant direct methods

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.47 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.45 \text{ e } \text{\AA}^{-3}$

Extinction correction: none

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cu1	0.53837 (2)	0.371184 (17)	0.64815 (2)	0.01513 (8)
O1	0.35686 (13)	0.37764 (9)	0.64009 (13)	0.0196 (3)
O2	0.27817 (13)	0.40992 (10)	0.43389 (12)	0.0198 (3)
O3	-0.35170 (13)	0.37559 (10)	0.45053 (13)	0.0222 (3)
O4	-0.27850 (13)	0.36803 (9)	0.66048 (13)	0.0192 (3)
O5	0.59539 (14)	0.34997 (10)	0.86385 (13)	0.0250 (3)
O1w	0.55072 (14)	0.50852 (10)	0.65718 (13)	0.0183 (3)
H1w1	0.6064 (19)	0.5313 (18)	0.629 (2)	0.050 (9)*
H1w2	0.4812 (15)	0.5389 (15)	0.625 (2)	0.037 (7)*
N1	0.52007 (16)	0.23600 (12)	0.59914 (15)	0.0169 (4)
N2	0.5139 (2)	0.37178 (13)	1.02550 (16)	0.0292 (4)
C1	0.6080 (2)	0.17317 (15)	0.66171 (18)	0.0218 (5)
H1	0.6770	0.1931	0.7336	0.026*
C2	0.6031 (2)	0.08068 (15)	0.62679 (18)	0.0227 (5)
H2	0.6676	0.0384	0.6747	0.027*
C3	0.50411 (19)	0.04921 (14)	0.52162 (17)	0.0165 (4)
C4	0.4123 (2)	0.11503 (14)	0.45831 (19)	0.0232 (5)
H4	0.3416	0.0972	0.3866	0.028*
C5	0.4237 (2)	0.20588 (15)	0.49936 (19)	0.0239 (5)
H5	0.3596	0.2494	0.4543	0.029*
C6	0.26399 (19)	0.39180 (13)	0.53743 (18)	0.0164 (4)
C7	0.12944 (18)	0.38330 (13)	0.54406 (18)	0.0158 (4)
C8	0.02418 (19)	0.39287 (14)	0.43468 (18)	0.0188 (4)
H8	0.0387	0.4034	0.3565	0.023*
C9	-0.10098 (19)	0.38726 (14)	0.43854 (18)	0.0196 (4)
H9	-0.1722	0.3932	0.3628	0.024*
C10	-0.12433 (19)	0.37297 (13)	0.55176 (18)	0.0164 (4)
C11	-0.0189 (2)	0.36227 (15)	0.66153 (19)	0.0238 (5)
H11	-0.0335	0.3517	0.7397	0.029*
C12	0.1059 (2)	0.36693 (15)	0.65759 (19)	0.0231 (5)
H12	0.1771	0.3589	0.7330	0.028*
C13	-0.26200 (19)	0.37214 (13)	0.55229 (19)	0.0173 (4)
C14	0.5048 (2)	0.35112 (15)	0.90772 (19)	0.0236 (5)
H14	0.4208	0.3358	0.8520	0.028*
C15	0.4001 (3)	0.37555 (17)	1.0660 (2)	0.0390 (6)
H15A	0.3244	0.3522	0.9984	0.058*
H15B	0.3844	0.4401	1.0856	0.058*

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H15C	0.4145	0.3368	1.1414	0.058*
C16	0.6383 (3)	0.3923 (2)	1.1183 (2)	0.0481 (7)
H16A	0.7072	0.3867	1.0796	0.072*
H16B	0.6548	0.3483	1.1887	0.072*
H16C	0.6373	0.4560	1.1497	0.072*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cu1	0.00697 (13)	0.02238 (14)	0.01633 (13)	-0.00036 (10)	0.00415 (9)	-0.00054 (10)
O1	0.0070 (7)	0.0310 (8)	0.0204 (7)	-0.0001 (6)	0.0038 (6)	-0.0001 (6)
O2	0.0123 (7)	0.0279 (8)	0.0207 (7)	-0.0020 (6)	0.0076 (6)	0.0006 (6)
O3	0.0092 (7)	0.0330 (9)	0.0233 (7)	0.0020 (6)	0.0037 (6)	0.0009 (6)
O4	0.0108 (7)	0.0275 (8)	0.0207 (7)	-0.0001 (6)	0.0070 (6)	0.0002 (6)
O5	0.0178 (8)	0.0371 (9)	0.0197 (7)	-0.0011 (7)	0.0054 (6)	0.0013 (6)
O1w	0.0109 (8)	0.0234 (8)	0.0219 (7)	-0.0002 (6)	0.0069 (6)	0.0013 (6)
N1	0.0108 (8)	0.0241 (9)	0.0161 (8)	0.0004 (7)	0.0046 (6)	0.0004 (7)
N2	0.0340 (12)	0.0355 (11)	0.0175 (9)	0.0008 (9)	0.0075 (8)	-0.0031 (8)
C1	0.0176 (11)	0.0270 (12)	0.0175 (10)	-0.0003 (9)	0.0008 (8)	0.0003 (9)
C2	0.0187 (11)	0.0243 (12)	0.0200 (10)	0.0020 (9)	-0.0008 (9)	0.0021 (9)
C3	0.0125 (10)	0.0247 (11)	0.0140 (9)	0.0000 (8)	0.0063 (8)	0.0028 (8)
C4	0.0154 (10)	0.0266 (12)	0.0217 (10)	0.0023 (9)	-0.0024 (9)	-0.0030 (9)
C5	0.0158 (11)	0.0256 (12)	0.0249 (11)	0.0040 (9)	-0.0010 (9)	-0.0013 (9)
C6	0.0122 (10)	0.0158 (10)	0.0224 (10)	-0.0008 (8)	0.0073 (8)	-0.0017 (8)
C7	0.0097 (9)	0.0183 (10)	0.0206 (10)	0.0008 (8)	0.0066 (8)	0.0010 (8)
C8	0.0147 (10)	0.0259 (11)	0.0174 (9)	-0.0005 (9)	0.0075 (8)	-0.0012 (8)
C9	0.0096 (9)	0.0283 (12)	0.0192 (10)	0.0012 (8)	0.0023 (8)	-0.0022 (8)
C10	0.0089 (9)	0.0180 (10)	0.0238 (10)	-0.0001 (8)	0.0075 (8)	-0.0007 (8)
C11	0.0140 (11)	0.0383 (13)	0.0212 (10)	0.0021 (9)	0.0088 (8)	0.0067 (9)
C12	0.0109 (10)	0.0372 (13)	0.0199 (10)	0.0016 (9)	0.0031 (8)	0.0043 (9)
C13	0.0105 (9)	0.0169 (10)	0.0259 (10)	-0.0001 (8)	0.0077 (8)	-0.0005 (8)
C14	0.0251 (12)	0.0275 (12)	0.0159 (10)	-0.0003 (9)	0.0036 (9)	0.0012 (8)
C15	0.0495 (17)	0.0443 (16)	0.0333 (13)	0.0032 (13)	0.0275 (12)	-0.0036 (11)
C16	0.0508 (18)	0.0621 (19)	0.0228 (12)	-0.0050 (14)	-0.0001 (12)	-0.0094 (12)

Geometric parameters (\AA , $^\circ$)

Cu1—O1	1.943 (1)	C3—C3 ⁱⁱⁱ	1.486 (4)
Cu1—O4 ⁱ	1.946 (1)	C4—C5	1.374 (3)
Cu1—O5	2.312 (1)	C4—H4	0.950
Cu1—O1w	1.975 (2)	C5—H5	0.950
Cu1—N1	2.008 (2)	C6—C7	1.489 (3)
O1—C6	1.277 (2)	C7—C8	1.387 (3)
O2—C6	1.245 (2)	C7—C12	1.394 (3)
O3—C13	1.238 (2)	C8—C9	1.374 (3)
O4—C13	1.281 (2)	C8—H8	0.950
O4—Cu1 ⁱⁱ	1.9462 (14)	C9—C10	1.385 (3)
O5—C14	1.231 (3)	C9—H9	0.950

O1w—H1w1	0.83 (1)	C10—C11	1.392 (3)
O1w—H1w2	0.84 (1)	C10—C13	1.495 (3)
N1—C5	1.333 (3)	C11—C12	1.369 (3)
N1—C1	1.335 (3)	C11—H11	0.950
N2—C14	1.324 (3)	C12—H12	0.950
N2—C16	1.444 (3)	C14—H14	0.950
N2—C15	1.445 (3)	C15—H15A	0.980
C1—C2	1.379 (3)	C15—H15B	0.980
C1—H1	0.950	C15—H15C	0.980
C2—C3	1.389 (3)	C16—H16A	0.980
C2—H2	0.950	C16—H16B	0.980
C3—C4	1.390 (3)	C16—H16C	0.980
O1—Cu1—O4 ⁱ	178.07 (6)	O2—C6—C7	118.66 (17)
O1—Cu1—O1w	90.33 (6)	O1—C6—C7	116.35 (17)
O4 ⁱ —Cu1—O1w	88.21 (6)	C8—C7—C12	118.79 (18)
O1—Cu1—N1	91.36 (6)	C8—C7—C6	119.31 (17)
O4 ⁱ —Cu1—N1	90.34 (6)	C12—C7—C6	121.90 (17)
O1w—Cu1—N1	167.58 (6)	C9—C8—C7	120.50 (18)
O1—Cu1—O5	88.96 (6)	C9—C8—H8	119.8
O4 ⁱ —Cu1—O5	89.91 (5)	C7—C8—H8	119.8
O1w—Cu1—O5	94.98 (5)	C8—C9—C10	120.69 (18)
N1—Cu1—O5	97.35 (6)	C8—C9—H9	119.7
C6—O1—Cu1	122.90 (13)	C10—C9—H9	119.7
C13—O4—Cu1 ⁱⁱ	112.40 (12)	C9—C10—C11	118.94 (18)
C14—O5—Cu1	115.77 (13)	C9—C10—C13	118.77 (17)
Cu1—O1w—H1w1	114.6 (19)	C11—C10—C13	122.27 (18)
Cu1—O1w—H1w2	117.2 (17)	C12—C11—C10	120.41 (19)
H1w1—O1w—H1w2	107 (3)	C12—C11—H11	119.8
C5—N1—C1	117.13 (18)	C10—C11—H11	119.8
C5—N1—Cu1	121.62 (14)	C11—C12—C7	120.65 (19)
C1—N1—Cu1	121.18 (14)	C11—C12—H12	119.7
C14—N2—C16	121.0 (2)	C7—C12—H12	119.7
C14—N2—C15	121.5 (2)	O3—C13—O4	124.35 (18)
C16—N2—C15	117.4 (2)	O3—C13—C10	119.09 (18)
N1—C1—C2	123.03 (18)	O4—C13—C10	116.56 (17)
N1—C1—H1	118.5	O5—C14—N2	125.9 (2)
C2—C1—H1	118.5	O5—C14—H14	117.0
C1—C2—C3	120.10 (19)	N2—C14—H14	117.0
C1—C2—H2	120.0	N2—C15—H15A	109.5
C3—C2—H2	120.0	N2—C15—H15B	109.5
C2—C3—C4	116.30 (19)	H15A—C15—H15B	109.5
C2—C3—C3 ⁱⁱⁱ	122.3 (2)	N2—C15—H15C	109.5
C4—C3—C3 ⁱⁱⁱ	121.4 (2)	H15A—C15—H15C	109.5
C5—C4—C3	120.08 (18)	H15B—C15—H15C	109.5
C5—C4—H4	120.0	N2—C16—H16A	109.5
C3—C4—H4	120.0	N2—C16—H16B	109.5
N1—C5—C4	123.35 (19)	H16A—C16—H16B	109.5

supplementary materials

N1—C5—H5	118.3	N2—C16—H16C	109.5
C4—C5—H5	118.3	H16A—C16—H16C	109.5
O2—C6—O1	124.98 (18)	H16B—C16—H16C	109.5
O1w—Cu1—O1—C6	83.52 (14)	Cu1—O1—C6—O2	-5.7 (3)
N1—Cu1—O1—C6	-84.17 (15)	Cu1—O1—C6—C7	172.75 (12)
O5—Cu1—O1—C6	178.50 (14)	O2—C6—C7—C8	2.0 (3)
O1—Cu1—O5—C14	-1.64 (15)	O1—C6—C7—C8	-176.58 (17)
O4 ⁱ —Cu1—O5—C14	176.79 (15)	O2—C6—C7—C12	-177.32 (19)
O1w—Cu1—O5—C14	88.60 (15)	O1—C6—C7—C12	4.1 (3)
N1—Cu1—O5—C14	-92.88 (15)	C12—C7—C8—C9	0.7 (3)
O1—Cu1—N1—C5	50.14 (17)	C6—C7—C8—C9	-178.65 (18)
O4 ⁱ —Cu1—N1—C5	-130.78 (17)	C7—C8—C9—C10	0.7 (3)
O1w—Cu1—N1—C5	-47.6 (4)	C8—C9—C10—C11	-1.4 (3)
O5—Cu1—N1—C5	139.27 (16)	C8—C9—C10—C13	176.95 (18)
O1—Cu1—N1—C1	-133.23 (16)	C9—C10—C11—C12	0.8 (3)
O4 ⁱ —Cu1—N1—C1	45.85 (16)	C13—C10—C11—C12	-177.54 (19)
O1w—Cu1—N1—C1	129.0 (3)	C10—C11—C12—C7	0.6 (3)
O5—Cu1—N1—C1	-44.10 (16)	C8—C7—C12—C11	-1.3 (3)
C5—N1—C1—C2	0.6 (3)	C6—C7—C12—C11	177.98 (19)
Cu1—N1—C1—C2	-176.20 (16)	Cu1 ⁱⁱ —O4—C13—O3	-1.6 (2)
N1—C1—C2—C3	0.5 (3)	Cu1 ⁱⁱ —O4—C13—C10	178.28 (12)
C1—C2—C3—C4	-1.3 (3)	C9—C10—C13—O3	6.6 (3)
C1—C2—C3—C3 ⁱⁱⁱ	178.7 (2)	C11—C10—C13—O3	-175.09 (19)
C2—C3—C4—C5	1.1 (3)	C9—C10—C13—O4	-173.25 (17)
C3 ⁱⁱⁱ —C3—C4—C5	-178.9 (2)	C11—C10—C13—O4	5.1 (3)
C1—N1—C5—C4	-0.8 (3)	Cu1—O5—C14—N2	-155.75 (18)
Cu1—N1—C5—C4	175.95 (16)	C16—N2—C14—O5	-2.5 (3)
C3—C4—C5—N1	0.0 (3)	C15—N2—C14—O5	176.9 (2)

Symmetry codes: (i) $x+1, y, z$; (ii) $x-1, y, z$; (iii) $-x+1, -y, -z+1$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1w—H1w1 \cdots O2 ^{iv}	0.83 (1)	1.83 (1)	2.658 (2)	175 (3)
O1w—H1w2 \cdots O3 ^v	0.84 (1)	1.85 (1)	2.684 (2)	168 (3)

Symmetry codes: (iv) $-x+1, -y+1, -z+1$; (v) $-x, -y+1, -z+1$.

Fig. 1

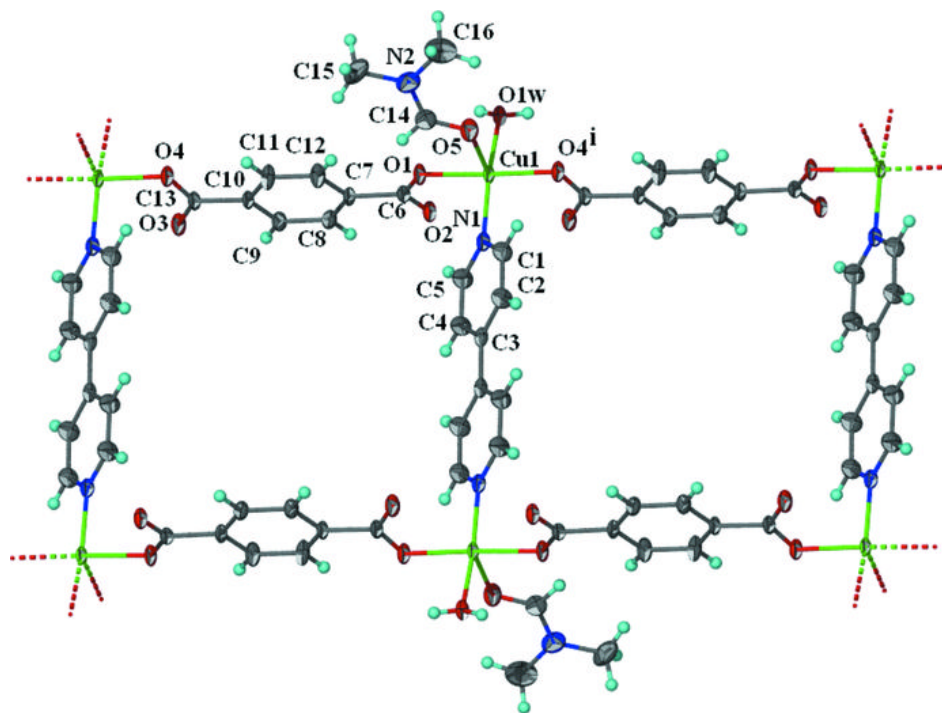


Fig. 2

